



Management of Phenols and Microoxygenation

Enology-Grape Chemistry Group

Virginia Tech

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Presentation Goals

- Nature of grape and wine phenols
- Chemistry of tannin-anthocyanin polymer formation
- Our research on microoxygenation

Grape and Wine Color

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Structure

- Sources: Skin, seeds, pulp, stems, oak, addition products
- Association with other compounds, macromolecules
- Size or degree of polymerization (DP)
- Sensory features: Astringent / Bitter / Color

Tannin Perception

Glycerol

Acidity

Manoproteins

Sugar — **Tannin** — Source/Structure

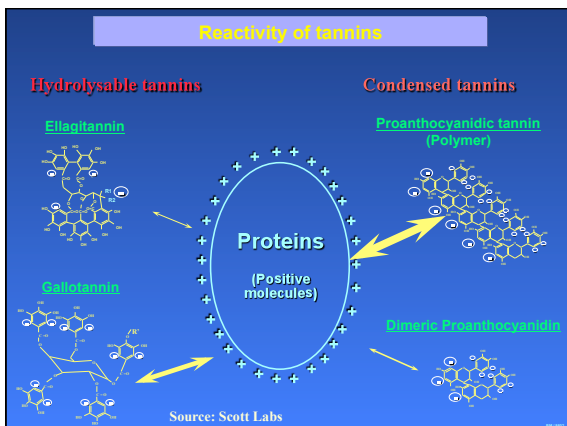
Ethanol

Anthocyanins

Pectin

Redistributing the DP of the monomers, and polymers

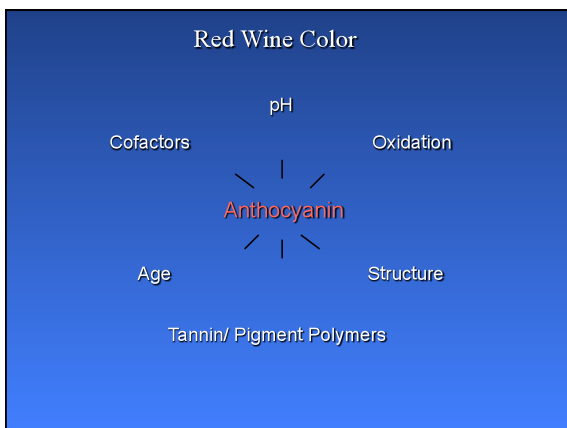
Breakage and reformation at (4→8) and (4→6) bonds



Nature of Astringency

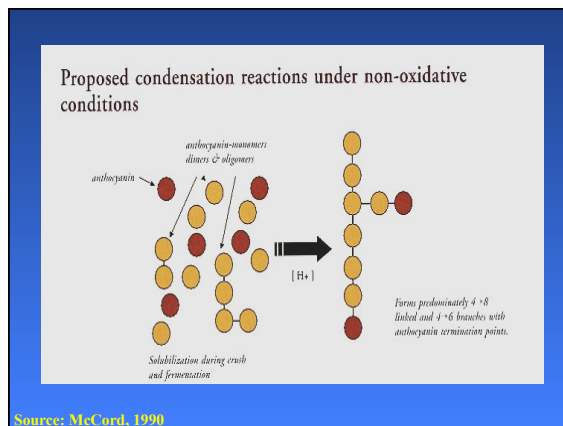
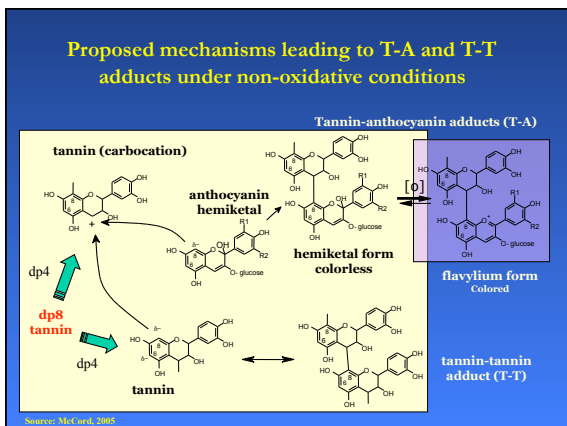
- Stereo-specific nature
- Size
- Number of OH

- Skin tannin DP \approx 40
finer grain of coarseness, not as lingering
- Seed and stem tannin DP \approx 15



Grape and Wine Pigment

Monomeric Anthocyanins (MP)	Red
Small Polymeric Pigments (SPP)	Red / Brown
Large Polymeric Pigments (LPP)	Red / Brown





Crosslinking with aldehydes

2

Acetaldehyde (or any aldehyde)*

Crosslinked produced polymers are formed with different bonds therefore different structures.

(8→8)

(6→8)

(6→6)

Acetaldehyde Production in Wine

Oxygen exposure (microox)

$$R-C_6H_3(OH)_2 + O_2 \rightarrow R-C_6H_3(O)_2 + H_2O_2$$

Heavy metal attack

$$R-C_6H_3(OH)_2 + 2Fe^{2+} \rightarrow R-C_6H_3(O)_2 + 2H^+ + 2Fe^{3+}$$

$$H_2O_2 + Fe^{2+} \rightarrow Fe^{3+} + OH^- + OH^\cdot$$

$$OH^\cdot + CH_3CH_2OH \rightarrow H_2O + CH_3CHOH^\cdot$$

$$CH_3CHOH^\cdot + Fe^{2+} \rightarrow CH_3CHO + H^+ + Fe^{2+}$$

$$H_2O_2 + CH_3CH_2OH \rightarrow CH_3CHO + 2H_2O$$

Aldehyde Compounds Extracted from Oak Capable Of Crosslinking Tannins and Anthocyanins

Acetaldehyde

Furfural

Coniferaldehyde

HMF (hydroxymethyl-furfural)

Cyclotene

Proposed condensation reactions under oxidative conditions

anthocyanin

anthocyanin-monomers dimers & oligomers

Accessible furfural, HMF, 5-MF

Solubilization during crush and fermentation with acetate

Form predominantly 8-8, 6+8 and 6+6

Inhib oligomers with anthocyanin termination points

Source: McCord, 1990

Phenol Relationships

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- Polymeric pigment formation impacts color, color stability, and mouthfeel
- Tannin and anthocyanin concentration effects concentration of polymers like LPP
- Low concentration of tannin – cannot form polymers
- Whichever is limiting will determine amount of polymeric pigment
- Stability of color is best when the ratio of anthocyanin ; tannin is 1 : 4
- What are viticultural and winemaking impacts?

Spectral Color

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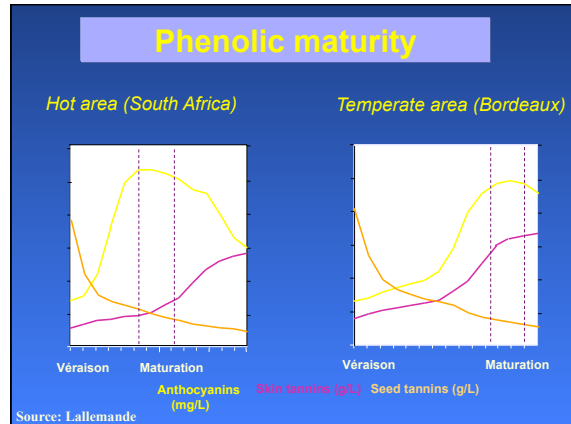
- Anthocyanin concentration
- Polymeric anthocyanins
 - SPP (small polymeric pigments)
 - LPP (large polymeric pigments)
- Cofactor concentration
 - Copigmentation can provide 4 to 10X more color
 - Non-Flavonoids
 - Flavonols
 - Arginine
 - Enological tannins? – NO



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Spectral Color

- Variation in cofactor concentration more important than anthocyanin
- Lots of seasonal variation
- Large cultivar differences



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Grape and Wine Color

- Change in grape anthocyanin concentration = Change in wine color
- Change in grape anthocyanin \neq Change in wine color
- No change in grape anthocyanin = Change in wine color

Pinot noir

- Skin tannin concentration X 6
- Seed tannin concentration X 2
- Yet 30 X difference in wine tannin concentration

Source: Adams, 2008

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Extraction Mainly a Function of:

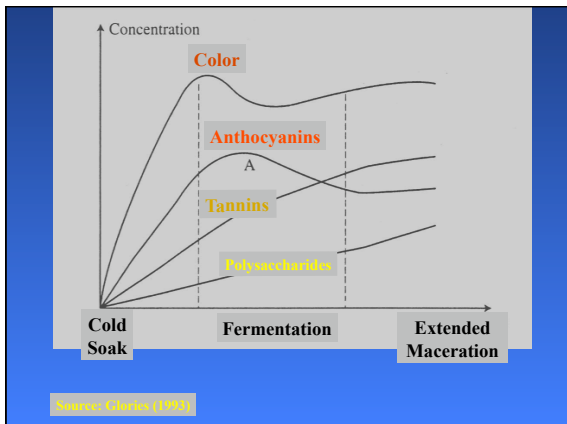
- Phenolic maturity – stems, cap stems, skins, seeds, pulp
- Temperature
- Contact
- Alcohol content

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Winemaking Techniques Influencing Phenols

Main Effects

- Concentration of phenolic elements
- Size of tannin (DP)
- Degree of galloylation
- Degree of trihydroxylation



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Cold Soak Variables

- Varieties
- Seasons
- Duration
- Temperature

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Cold Soak Effects

- Increased anthocyanin extraction
- Increased tannin extraction
- Change in A / T ratio
- Increase in spectral color?
- Increase in DP
- Increase in wine tannin

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Saignée

- How much?
- ↑ Tannins
- ↑ DP

Effect of Micro-Oxygenation on Wine Phenols and Sensory Profiles

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Microoxygenation Potential

- Improve wine mouthfeel
- Increase in volume or palate weight
- Color stability
- Ameliorate "green" herbaceous aromas
- Ameliorate SLO character



Microoxygenation Variables

- Quantitative and Qualitative nature of phenols
 - degree of polymerization
- Timing
- Concentration of oxygen
- Wine temperature
- Sulfur dioxide concentration
- The wine

Types of Microoxygenation

- **Micro-bullage**

- **Dense polymer membrane**

Sensory Evaluation

- 16 trained subjects
- Intensity scale 1-9
- Aroma Descriptors
 - fruit, veggy, spicy, oak, herbal, oxidation
- Taste Descriptors
 - fruit, veggy, green-tannin, tannin-grit, plushness, off-aroma

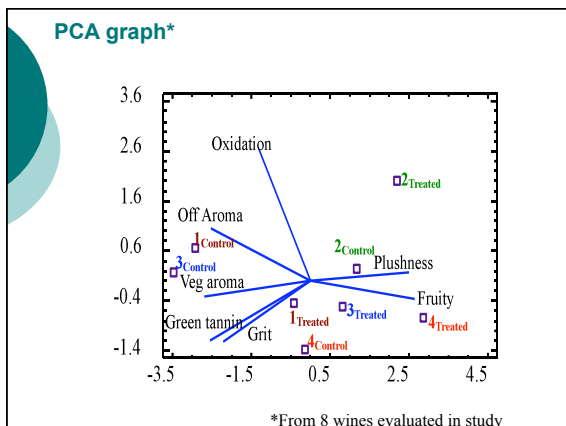
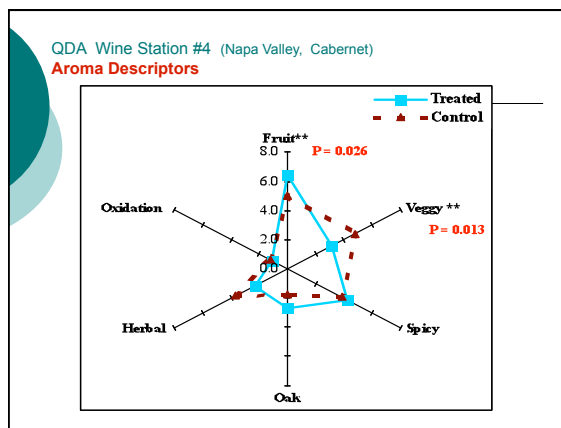
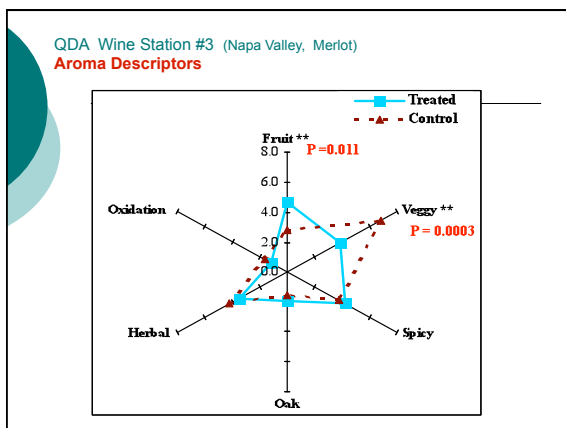
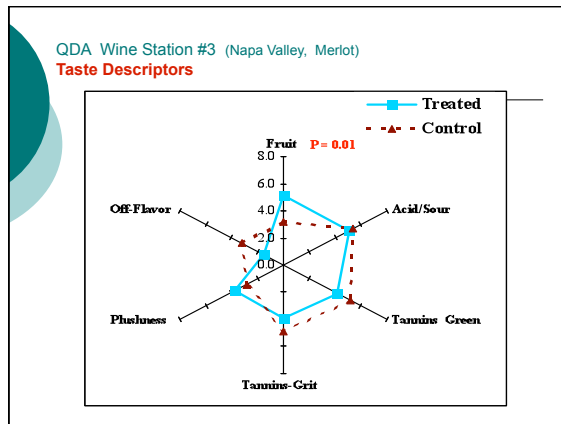
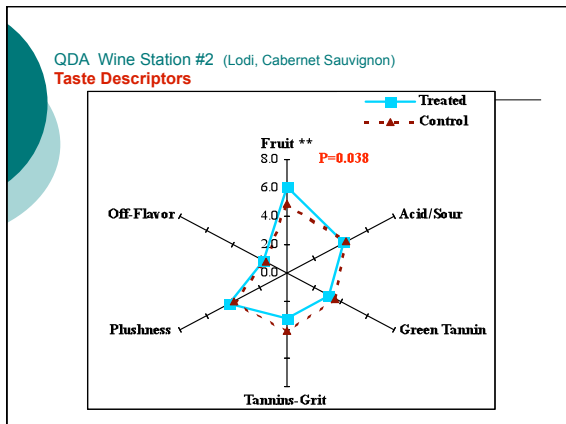
Tannin Development

Source: Zoecklein et al., 2003

Microoxygenation

Source: Zoecklein et al., 2003

Equilibrium Between Colored Malvidin-3-glucoside and its Colorless Bisulfite Addition Product



Correlation Coefficients

		R Value (Correlation Coefficient)						
		Fruit	Vegetative	Oxidation	Off aroma	Green Tannin	Tannin Grit	Plushness
P Value	Fruit		-0.938	-0.448	-0.958	-0.836	-0.590	0.784
	Vegetative	0.0006***		0.354	0.862	0.910	0.796	-0.782
	Oxidation	0.2661	0.3903		0.600	0.127	0.233	-0.389
	Off aroma	0.0002***	0.0059*	0.1184		0.711	0.471	-0.770
	Green Tannin	0.0097**	0.0017**	0.7642	0.0482*		0.841	-0.824
	Tannin-Grit	0.1237	0.018*	0.5783	0.2394	0.0088*		-0.721
	Plushness	0.0212*	0.0218*	0.3410	0.0253*	0.0119*	0.0436*	



Phenolic Results

- 8 different wines.
- The average results showed the following trends:
 - Increases in visible color
 - Increases in color intensity
 - Increases in polymeric color
 - Increases in browning
 - Increases in age related factors

Microoxygenation Issues

- **Timing and Concentration of Oxygen**
- **Concentration of pigment polymers**
- **Structural phase**
- **Harmonization phase**
- **Sulfur dioxide concentration**



Much more info on phenols and microoxygenation at
www.vtwines.info