

1 **Phosphorus availability from bone char in a P-fixing soil influenced by root-mycorrhizae-**
2 **biochar interactions**

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30 **Abstract**

31 *Aims*

32 The objectives of this study were to evaluate (1) the fertilizer potential of bone char, (2) the
33 effects of wood biochar on plant-available phosphorus (P), and (3) the role of root-mycorrhizae-
34 biochar interactions in plant P acquisition from a P-fixing soil.

35 *Methods*

36 Incubation and pot experiments were conducted with a P-fixing soil and maize with or without
37 root hairs and arbuscular mycorrhizae (AM) associations. Olsen-, resin-P and plant P
38 accumulation were used to estimate P availability from bone char, co-pyrolyzed bone char-wood
39 biochar, and separate bone char and wood biochar additions produced at 60, 350 and 750°C, and
40 Triple Superphosphate (TSP).

41 *Results*

42 Maize inoculated with AM showed similar P accumulation when fertilized with either 750°C
43 bone char or TSP. Pyrolyzing bone did not increase extractable P in soil in comparison to
44 unpyrolyzed bone, apart from a 67% increase in resin-extractable P after additions of bone char
45 pyrolyzed at 350°C. Despite greater Olsen-P extractability, co-pyrolysis of bone with wood
46 reduced maize P uptake. Wood biochars reduced resin-P from bone char by 14-26%, whereas
47 oven-dried wood increased resin-P by 23%.

48 *Conclusions*

49 Bone char is an effective P fertilizer, especially if root-AM interactions are simultaneously
50 considered. Biochar influences plant access to soil P and requires careful management to
51 improve P availability.

- 52 **Keywords:** Arbuscular mycorrhizae; biochar; bone char; nutrient acquisition strategies;
- 53 phosphorus adsorption

54 **Introduction**

55 Managing soil phosphorus (P) availability is essential to sustain agricultural production
56 worldwide. Phosphorus adsorption to mineral oxides results in soil fertility management
57 challenges, especially in highly weathered acid soils (Sanchez 1976; Marschner 1995; Haynes
58 and Mokolobate 2001; Hinsinger 2001; Raghothama and Karthikeyan 2005). In addition, rock
59 phosphate mining for fertilizer is rapidly depleting global P reserves (Abelson 1999; Smil 2000;
60 Gilbert et al. 2009). Hence, there is an urgent need for developing alternative P fertilizers and
61 sustainable management practices that optimize P availability in agro-ecosystems (Cordell et al.
62 2009, 2011).

63 Converting slaughterhouse waste into P fertilizer through pyrolysis is one way to recycle
64 P to agricultural systems (Vassilev et al. 2013). Not only would this renewable fertilizer be a
65 more economic option (Buss et al. 2016), bone chars contain fewer heavy metals and are more
66 effective in sorbing cadmium from contaminated soils than conventional fertilizer (Siebers and
67 Leinweber 2013). Animal bones mainly consist of biological apatite, a relatively crystalline
68 calcium phosphate structure (Wopenka and Pasteris 2003, 2005). In contrast to pyrolyzed plant
69 materials or biochar, pyrolyzed bone contains low amounts of organic carbon. Therefore, it is not
70 referred to as biochar but bone char. While bone char as sustainable P fertilizer has received
71 more scientific attention over the past few years, little remains known about the effect of
72 pyrolysis conditions on the P fertilizer efficacy of bone char. Moreover, the biological and
73 chemical mechanisms that control plant P availability from bone char in soils are poorly
74 understood.

75 Previous studies found that bone char produced at 400°C was a more effective P fertilizer
76 than GAFSA rock phosphate (Warren et al. 2009). Research focusing on the effect of production

77 conditions on availability of bone char P showed that pyrolysis temperature decreases water-
78 extractable P but increases formic acid-extractable and presumably plant-available P (Zwetsloot
79 et al. 2015). Yet it is unclear how these results translate to actual P uptake of plants. In fact,
80 greenhouse experiments demonstrated that bone char application to different soils and crops led
81 to both increases and decreases in plant biomass and P concentration (Siebers et al. 2012, 2014).
82 These findings warrant further investigation of the availability of bone char P resulting from
83 different production conditions in order to compare bone char to conventional P fertilizers such
84 as Triple Super Phosphate (TSP) fertilizer.

85 Besides testing the P fertilizer efficacy of different bone chars, managing organic matter
86 inputs and root-mycorrhizae interactions could optimize P availability from bone char. In
87 comparison to other nutrients, P has low mobility in soils. Especially in P-fixing soils commonly
88 found in tropical farming systems (Sanchez 1976), plant-available P from fertilizer can be
89 drastically reduced through its interactions with mineral oxides that chemisorb phosphate from
90 the soil solution (Parfitt et al. 1975). Organic matter additions to P-fixing soils can improve the
91 availability of added P by decreasing P adsorption and increasing P desorption (Singh and Jones
92 1976; Guppy et al. 2005). The same has been observed for biochar additions to ferrihydrite
93 minerals (Cui et al. 2011). The presence of biochar in Ferralsol and Anthrosol soils increased
94 total plant P uptake when P fertilizer was added (Lehmann et al. 2003). Yet other studies found
95 that biochar only decreased P adsorption in a Ferralsol when compost was added simultaneously
96 (Qayyum et al. 2015). It is unclear what biochar production parameters and plant growing
97 conditions may explain these different results.

98 Recent research has highlighted the importance of root foraging to plant P nutrition
99 (Ramaekers et al. 2010; Richardson et al. 2011). Increased root hair elongation, top soil lateral

100 branching, high root:shoot ratio, root exudation, increased root P uptake kinetics and mycorrhizal
101 symbiosis are all found to be strategies that favor P acquisition (Lynch and Beebe 1995; Bates
102 and Lynch 2001; Vance et al. 2003; Hodge et al. 2009; Ramaekers et al. 2010; Zhu et al. 2010;
103 Lynch 2011). Yet few studies have simultaneously examined the effect of soil amendments and
104 nutrient acquisition strategies on plant P uptake. Especially in the case of P fertilizers with low
105 water-solubility such as bone char, managing the foraging capacity of plants and mycorrhizal
106 associations may significantly improve plant P acquisition. Yet we are not aware of studies
107 examining the combined effects of plant rooting or mycorrhizae and P uptake from bone char.

108 Therefore, the objectives of this study were: (1) to test the potential of bone char
109 produced at different pyrolysis temperatures as a P fertilizer in comparison to TSP; (2) to
110 examine the effect of co-pyrolyzed and separate maple wood biochar additions on P availability
111 from bone char in a P-fixing soil; and (3) to evaluate the role of AM and root foraging strategies
112 in P acquisition from bone char and biochar applications.

113

114 **Materials and Methods**

115 *Soil Characteristics*

116 Soil was collected from the Jimma University agricultural research farm (Jimma, Ethiopia).
117 Mehlich-III extractable P was below our detection limit of 0.01 mg P kg⁻¹ soil. Soil pH was 4.9 in
118 water and 4.0 in 1 M KCL. The cation exchange capacity was 279 mmol_c kg⁻¹ soil, and it
119 contained exchangeable base cations at levels of 5.2 g Ca kg⁻¹ soil, 2.9 g K kg⁻¹ soil, 1.2 g Mg
120 kg⁻¹ soil and 0.1 g Na kg⁻¹ soil. Total C was 31 g kg⁻¹ and total N 3.0 g kg⁻¹. The soil showed a
121 substantial maximum P sorption capacity of 456 mg P kg⁻¹ soil (Langmuir isotherm with k =
122 0.171, R² = 0.98, n=9). Soil texture was 50% clay, 40% silt and 10% sand as determined by

123 hydrometer analysis. The clay fraction mainly contained kaolinites with traces of illite, chlorite
124 and quartz as shown by X-ray diffraction (Online Supplementary Fig. S1).

125

126 *Phosphorus Fertilizers and Biochars*

127 Rendered bone meal was purchased from The Espoma Company 1929 (Milville, NJ, USA).

128 Robinson Lumber in Owego, NY, USA supplied hard wood chips (80% red maple, 20% sugar
129 maple). Wood chips were oven-dried at 60°C and ground to a particle size < 2 mm with a

130 Thomas Wiley Mill (Thomas Scientific, Swedesboro, NJ, USA). Bone meal was pyrolyzed at a

131 heating rate of 2.5°C min⁻¹ and temperature was maintained at either 350 or 750°C for 45 min in

132 a muffle furnace swept with argon gas. Bone meal mixed with hard wood chips at a dry weight

133 ratio of 1:1 were pyrolyzed under the same conditions. Greenkeeper's Secret Triple

134 Superphosphate (TSP) fertilizer (T&N, Inc., Foristell, MO, USA) supplied soluble P in the

135 fertilizer controls. The pH, formic- and water-extractable P, total elemental P, Ca, Mg, K, Na, S,

136 Fe and Al (Zwetsloot et al. 2015), and total elemental C, N, O and H by combustion using a

137 Thermo Delta V Advantage Isotope Ratio Mass Spectrometer and a Temperature Conversion

138 Elemental Analyzer (Thermo Scientific, West Palm Beach, FL) were measured to characterize the

139 bone chars, biochars and fertilizers (Table 1).

140

141 *Abiotic Incubation Design*

142 An abiotic incubation experiment compared 12 different P additions to soils in an incomplete

143 factorial design, by varying feedstock for pyrolysis, P source-biochar mixtures and pyrolysis

144 temperatures: (a) bone meal; (b) bone meal pyrolyzed at 350°C or 750°C; (c) mixtures of

145 separately oven-dried (60°C) and pyrolyzed (at 350°C or 750°C) bone meal and maple wood; (d)

146 co-pyrolyzed (at 350°C or 750°C) bone meal and wood; (e) TSP alone; and (f) TSP mixed
147 separately to soil with 350°C or 750°C wood biochar (Supplementary Online Table S1). These
148 were compared to unamended controls and to added wood that was either oven-dried at 60°C or
149 pyrolyzed at either 350°C or 750°C. Pyrolyzed materials were ground with a mortar and pestle to
150 a particle size of 74-150 µm. Bone meal and dried wood chips were ground with a Thomas
151 Wiley Mill (Thomas Scientific, Swedesboro, NJ, USA) to mesh size 60.

152 The treatments were applied in five replicates to 250-mL mason jars containing 5 g of
153 soil sieved to < 2 mm. Deionized water was added to saturation (4 mL jar⁻¹). To create abiotic
154 conditions for P sorption, HgCl₂ was dissolved in the deionized water prior to application at a
155 rate of 500 mg HgCl₂ kg⁻¹ soil (Tuominen et al. 1994). Bone and TSP treatments were applied at
156 a rate of 360 mg total P kg⁻¹ soil, which was based on 80% of the maximum sorption capacity of
157 soil. The separate maple wood biochar additions were equal to the amount of biochar applied in
158 the co-pyrolyzed bone char-wood biochar treatments amounting to a wood application rate of 4.2
159 g kg⁻¹, 1.2 g kg⁻¹ and 0.9 g kg⁻¹ soil for 60°C, 350°C and 750°C treatments respectively. Total P
160 content of wood biochars was less than 0.3 mg P g⁻¹ (Table 1), confirming that the wood biochars
161 by themselves were not a significant P nutrient source. All jars were put on a rotary shaker at 25
162 rpm at 30°C for five weeks.

163

164 *Abiotic Incubation Analyses*

165 Total contents of the jars (5 g) were used for analysis. Olsen-P and resin-P extractions were used
166 to measure plant-available P since more common extraction methods for low pH soils such as
167 Mehlich and Bray would have overestimated P from the calcium phosphate chemical structures
168 of bone char (Kuo 1996). In the case of Olsen-P, 100 mL of 0.5 M NaHCO₃ adjusted to pH 8.5

169 with 1 M NaOH solution was added to the samples and put on a horizontal shaker at 100 rpm for
170 30 min (Kuo 1996). After filtering with both a 2V Whatman qualitative filter paper and 0.45 μm
171 filter paper, the filtrates received 3.25 mL 12 M HCl to lower the pH for colorimetric P
172 determination (Murphy and Riley 1962).

173 For the resin-P extraction (Tiessen and Moir 1993), anion-exchange resin (AER) strips
174 (Ionac MA-7500, LANXESS Sybron, Birmingham, NJ, USA) were cut to 20 by 60 mm strips,
175 rinsed four times with deionized water, soaked in 0.5 M NaHCO_3 (25 strips L^{-1}) overnight, and
176 rinsed with deionized water. One AER strip together with 150 mL deionized water was added to
177 each jar and placed on a rotary shaker at 100 rpm for 24 hrs. AER strips were removed from the
178 jars and placed in centrifuge tubes containing 40 mL 0.5 M HCl. After shaking the centrifuge
179 tubes at 200 rpm for 20 hrs, AER strips were taken out and 0.5 M HCL extracts were
180 colorimetrically analyzed for P (Murphy and Riley 1962). Olsen-P and resin-P extracts from the
181 TSP treatments were analyzed by inductively coupled plasma atomic emission spectrometry
182 (ICP-AES Thermo Jarrel Ash 166 Trace Analyzer, Thermo Jarrell Ash Corporation, Franklin,
183 MA, USA).

184

185 *Pot Trial Design and Management*

186 The pot trial followed a completely randomized block design. Phosphorus source additions were
187 sieved to < 2 mm and included TSP fertilizer, bone char, and bone char-wood biochar
188 combination pyrolyzed at 750°C in comparison to a zero-P control. Because previous studies
189 showed higher formic-extractable and therefore presumably high plant-available P contents from
190 bone char with an increase in pyrolysis temperature (Zwetsloot et al. 2015), bone char produced
191 at 750°C was used for this trial. To examine the interaction of plant's P foraging ability with

192 these P sources, we both increased and decreased the P uptake efficiency of the single maize
193 variety B73 (*Zea mays* L.). Phosphorus foraging capacity was decreased by using a B73 mutant
194 with no root hairs (-RH) and increased by inoculating unmodified B73 with AM (+RH+AM).
195 The unmodified B73 variety was also planted (+RH). B73 maize seeds were supplied by the
196 USDA Genetic Resource Information Network (Beltsville, MD, USA). The root hairless rth1-1
197 mutant (115A) seeds (Wen and Schnable 1994, Hochholdinger et al. 2008) were obtained from
198 the Maize Genetics Coop Stock Center (Urbana, IL, USA). *Glomus clarum* (strain WV325,
199 INVAM, West Virginia University, USA) was used for AM inoculation. To ensure higher P
200 foraging ability of treatments inoculated with *G. clarum*, we measured AM colonization of maize
201 roots with trypan blue staining and microscopic analysis (Koske and Gemma 1989).

202 To stagger plant harvests, the five replicates of the experiment were planted on
203 successive days and placed in a 2.40 m by 1.35 m growth chamber at the Guterman Bioclimatic
204 Laboratory and Greenhouse Complex at Cornell University, Ithaca, NY, USA. Model CP512 tree
205 pots (volume = 5 L, height = 0.305 m, width = 0.127 m) were used (Stuewe and Sons, Tangent,
206 OR, USA). Pots were filled with 4 L of soil (dry weight = 3713 g) with a control or *G. clarum*
207 inoculum in the middle 1250 mL layer at a volume ratio of 1:25. Inoculant carrier material was
208 Agsorb attapulgite (Oildri, Chicago, IL, USA) with dried roots and spores from a low-P
209 sorghum-sudangrass pot culture. As a control, carrier material without inoculant was added to
210 treatments without AM. TSP fertilizer, bone char and co-pyrolyzed bone char-wood biochar
211 were added at the same total P application rate of 100 kg P ha⁻¹, which equaled to 43.4 mg P kg⁻¹
212 soil or 161.3 mg P pot⁻¹. The abiotic incubation used a very high P addition rate in order to study
213 chemical mechanisms controlling plant-available P. We lowered the application rate for the pot
214 trial to be more in line with field conditions. The wood biochar application in the bone char-

215 wood biochar treatment amounted to 0.01% by dry soil weight or 0.4 g biochar pot⁻¹. The total P
216 content of 750°C wood biochar was 0.28 mg P g⁻¹ biochar (Table 1). Hence, additional P supply
217 through biochar totaled 0.1 mg pot⁻¹ and was ruled out as relevant P source for maize. Three B73
218 maize seeds were placed 50 mm above the middle layer. Three days after germination, the two
219 weakest seedlings were removed. Treatments without root hairs only received one B73 115A
220 seed due to limited seed availability.

221 Pots were watered by weight to 50% water-filled pore space. Every other day plants were
222 watered with 100 mL of nutrient solution without P: 1.5 mM Ca(NO₃)₂, 5 mM NH₄NO₃, 3 mM
223 KNO₃, 2 mM (NH₄)₂SO₄, 2 mM MgSO₄, 0.024 mM Fe-EDTA, 0.05 mM KCL, 0.025 mM
224 H₃BO₄, 0.007 mM MnSO₄, 0.006 mM ZnSO₄, 0.002 mM CuSO₄ and 0.0005 mM (NH₄)₆Mo₇O₂₄.
225 On day 10, all pots received an additional 100 mL of 15 mM K₂SO₄.

226

227 *Pot Trial Analyses*

228 At harvest, five weeks after seeding, shoots were cut and pots were sliced vertically into two
229 halves. One half was carefully washed with water and sampled for intact crown roots that were
230 stained (0.167 g L⁻¹ neutral red dye) and used for morphological analyses. The remaining roots
231 were washed with water. The shoots and roots were dried at 60°C for 4 days and weighed. Dried
232 biomass was ground with a Thomas Wiley Mill mesh size 60 (Thomas Scientific, Swedesboro,
233 NJ, USA).

234 To determine plant P concentration and total P uptake, shoot and root biomass were digested
235 in a mixture of HNO₃ and 30% H₂O₂ (Benton Jones 2001). Digests were analyzed by inductively
236 coupled plasma atomic emission spectrometry (ICP-AES Thermo Jarrell Ash 166 Trace
237 Analyzer, Thermo Jarrell Ash Corporation, Franklin, MA).

238 Four replicates were used for root morphological analysis with WinRHIZO Pro 2007d
239 (Regent Instruments Inc., Québec, Canada). Red stained crown roots from harvest were scanned
240 and analyzed for number of first-order root tips (Supplementary Online Fig. S2).

241

242 *Statistical Analyses*

243 JMP Pro 10 was used for all statistical analyses (SAS, Cary, NC, USA). Student's t-tests were
244 used to determine differences between treatments. A linear regression between resin-P and
245 Olsen-P, total plant P uptake and resin-P, as well as total plant P uptake and Olsen-P established
246 the effectiveness of the abiotic incubation extraction methods to predict total plant P uptake.
247 Total plant biomass, plant P uptake, and plant calcium (Ca) uptake were transformed using a
248 natural logarithm before statistical analyses.

249

250 **Results**

251 *Abiotic Incubation*

252 Olsen-P extractions showed that increasing pyrolysis temperature decreased extractable P
253 ($p < 0.05$) from rendered bone (Fig. 1a). In comparison to bone char pyrolyzed alone at either
254 350°C and 750°C, co-pyrolyzing bone with wood at 350°C and 750°C led to a 13-19% increase
255 in Olsen-P ($p < 0.05$). Olsen-P values from unpyrolyzed (60°C) bone meal and bone meal
256 pyrolyzed at 350°C were 14-28% higher than Olsen-P from TSP ($p < 0.05$).

257 Resin-P and Olsen-P were only weakly correlated ($R^2 = 0.13$, $p < 0.05$). Resin-P was a
258 better predictor for maize P uptake than Olsen-P, with $R^2 = 0.80$ and $R^2 = 0.52$, respectively
259 (Supplementary Online Table S2). Pyrolyzing bone at 350°C resulted in 67% higher resin-P
260 compared to unpyrolyzed bone ($p < 0.05$, Fig. 1b); however, this effect disappeared at 750°C.

261 Both the separate addition of wood biochar or co-pyrolysis of bone and wood decreased resin-P
262 by 14-26% in comparison to bone char alone ($p<0.05$). Similar decreases in resin-P were
263 observed for all TSP and wood biochar treatments. Adding fresh biomass to the unpyrolyzed
264 bone led to a 23% increase in resin-P ($p<0.05$).

265

266 *Pot trial*

267 Plant P uptake by maize inoculated with AM was the same when receiving TSP or bone char
268 fertilizer (Fig. 2a, $p>0.05$), but significantly lower when fertilized with co-pyrolyzed bone char-
269 wood biochar ($p<0.05$). Without AM but with root hairs, TSP outperformed bone char fertilizer:
270 maize fertilized with either bone char or co-pyrolyzed bone char-wood biochar showed both
271 significantly lower P uptake than plants with TSP additions ($p<0.05$).

272 Plant Ca uptake followed similar trends as P uptake with two exceptions: the maize
273 without root hairs showed no significant differences among additions and AM inoculation did
274 not increase Ca uptake in co-pyrolyzed bone char-wood biochar or without additions (Fig. 2b,
275 $p>0.05$). Whole-plant P concentration was highest in maize inoculated with AM for all P
276 additions, while the reverse was true for Ca concentration (Supplementary Online Table S3).

277 The presence of root hairs and AM inoculation also increased total plant biomass
278 ($p<0.05$) for all P sources (Fig. 2c). Unlike trends observed for P uptake, TSP application
279 resulted in higher total plant biomass for maize with root hairs and maize with root hairs and AM
280 in comparison to other P additions ($p<0.05$). Maize with root hairs had higher total biomass
281 when fertilized with bone char than with co-pyrolyzed bone char-wood biochar ($p<0.05$), while
282 maize without root hairs did not respond to P additions ($p>0.05$).

283 The root:shoot ratios of maize without root hairs and with both root hairs and AM were
284 significantly higher than those of maize without root hairs ($p<0.05$, Fig. 2d). When maize was
285 inoculated with AM, the root:shoot ratios after bone char additions were greater than those after
286 TSP additions ($p<0.05$). Without P application, root:shoot ratios of maize with or without root
287 hairs were significantly lower than when other P sources were applied ($p<0.05$).

288 AM inoculation significantly increased AM colonization of roots ($p<0.05$) and did not
289 vary among different P additions (Fig. 3a). For maize with both root hairs and AM, the degree of
290 root branching measured as number of root tips was significantly higher for bone char and TSP
291 additions in comparison to co-pyrolyzed bone char-wood biochar and zero-P additions (Fig. 3b,
292 $p<0.05$). Diameter, average root length and surface area of first-order roots did not demonstrate
293 relevant trends among treatments (Fig. S3).

294

295 **Discussion**

296 *Bio-Availability of Phosphorus in Bone Char*

297 Bone char proved to be an effective P fertilizer in highly weathered, acid soils. When inoculated
298 with AM, maize plants acquired similar amounts of P from TSP and bone char pyrolyzed at
299 750°C. Both resin-P and Olsen-P from soil incubated with bone char produced at 350°C even
300 indicated a higher P availability than when soils were incubated with TSP. While the incubation
301 showed no difference in resin-P and Olsen-P between TSP and 750°C bone char, maize plants
302 without AM acquired more P from TSP in the pot trial. Unlike bone char, TSP is highly water-
303 soluble and therefore becomes less available to plant roots over time through adsorption
304 reactions with mineral oxide surfaces (Kucey et al. 1989). In the pot experiment, roots and AM
305 may have taken up soluble P before chemisorption to soil minerals, explaining the higher P

306 availability of TSP in comparison to measurements in soil incubation experiment. Given that
307 resin-P contents were higher when bones were pyrolyzed at 350°C than 750°C in the incubation
308 trial, P acquisition by maize from bone char may have been even higher if produced at lower
309 pyrolysis temperatures than 750°C bone char used in the pot trial. Since high soil acidity
310 facilitates the continued dissolution of apatite through the release of structural OH⁻ (Rajan et al.
311 1996), the effectiveness of bone char as fertilizer may be less in neutral or alkaline soil
312 environments than observed in the studied acid soil.

313 Greater resin-extractable P from bone char produced at 350°C in comparison to
314 unpyrolyzed bone meal additions to soil may be explained by the increase in hydroxyl apatite-
315 like crystals through pyrolysis, which reduces water-extractable P but increases formic-
316 extractable P (Zwetsloot et al. 2015), representing plant-available P. Hydroxyl apatite may be
317 more soluble than other calcium phosphates in solutions with low pH (Brown et al. 1975;
318 Matsumoto et al. 2002), improving P availability in acid soil environments. In addition, pyrolysis
319 between 250 and 500°C frees up the organic bone constituents (Deydier et al. 2005). Cleaving
320 organic P bonds could enhance extractable P (DeLuca et al. 2009). At pyrolysis temperatures
321 greater than 500°C, free P from organic bone constituents may become less available again
322 through its incorporation into hydroxyl apatite crystal lattice structures. When added to soil, bone
323 char produced at 750°C indeed results in lower resin-P than bone char produced at 350°C (Fig.
324 2b) despite its greater amounts of formic-acid extractable P reported previously (Zwetsloot et al.
325 2015).

326 The reduction in Olsen-P with an increase in pyrolysis temperature can also be explained
327 by an increase in crystallinity. While the OH⁻ anions from the Olsen solution (pH = 8.5) increase
328 P solubility through complexing or precipitating aluminum (Al³⁺) and iron (Fe²⁺), the increase in

329 soil solution pH might have limited the dissolution of calcium phosphate. Rather than the formic-
330 P fraction, Olsen-P relies more on the water-P fraction of bone char, which decreases with
331 greater calcium phosphate crystallization. Likewise, the increase in Olsen-P when rendered bone
332 is co-pyrolyzed with wood biomass is likely attributed to a decrease in crystal formation and
333 increase in water-P (Zwetsloot et al. 2015) rather than a reduction in soil P-fixation due to
334 biochar.

335

336 *Biochar Effects on P Availability*

337 Mixing maple wood biochar with a P source before or after pyrolysis did not increase P
338 availability in this study. This result was observed in both the abiotic incubation and pot
339 experiment, suggesting that largely chemical interactions between P source, biochar and soil
340 were responsible. The decrease in resin-P from TSP and bone fertilizer when mixed with biochar
341 additions may be explained by ash minerals from biochar precipitating P out of solution
342 (Hollister et al. 2013). In addition, biochar may have served as an adsorbent for organic acids
343 already present in the soil (Cornelissen et al. 2005) increasing P fixation to soil minerals. Low
344 molecular weight organic acids from root exudates and dead plant material are known to
345 decrease P-fixation through competition with P for adsorption surfaces or complexing with Fe^{2+}
346 and Al^{3+} , thereby increasing the P concentration in the soil solution (Bolan et al. 1994; Jones
347 1998; Hinsinger 2001; Antelo et al. 2007). In contrast to pyrolyzed biomass, adding oven-dried
348 wood biomass increased resin-P from bone meal possibly generating leachates that contain
349 additional low-molecular weight organic acids that reduced P-fixation as often shown for plant
350 residues (Earl et al. 1979; Haynes and Mokolobate 2001; Hunt et al. 2007). The importance of

351 this competitive adsorption of low-molecular weight acids to biochar for soil P availability
352 would need to be directly quantified in future experiments.

353 Managing soil pH is an important strategy for diminishing P adsorption to mineral
354 surfaces and may also influence the extent to which biochar changes P fixation (Cui et al. 2011).
355 Some biochars themselves can significantly increase soil pH thereby increasing the availability
356 of essential plant nutrients such as P (Van Zwieten et al. 2009; Atkinson et al. 2010). Yet a
357 significant biochar effect on P availability through a change in soil pH seems unlikely to explain
358 variations in plant-available P in this experiment. Soil pH in the pot trial was unaffected by
359 different bone char, TSP and biochar additions (Online Supplementary Table S4). While soil pH
360 significantly increased from 4.9 to 5.1 by bone meal and bone char additions in the abiotic
361 incubation (Online Supplementary Table S5), these changes did not correspond with higher
362 levels of resin-P ($r^2 = -0.19, p < 0.05$) and Olsen-P ($r^2 = 0.08, p > 0.05$). Moreover, wood
363 biochar by itself did not lead to a significant change in soil pH in comparison to the control
364 (Online Supplementary Table S5). Since the biochar application rates in this study were
365 relatively low, higher rates may need to be used in order to manage soil pH for enhancing plant-
366 available P.

367

368 *Root and AM exploitation of P sources*

369 Bone char presents a patchier P distribution in soil and takes longer to dissolve into solution than
370 TSP. This may explain why only maize inoculated with AM demonstrated similar P uptake
371 under TSP and bone char fertilizer additions. AM aid plants in foraging for unevenly distributed
372 nutrients by increasing the soil volume that is explored for P (Cui and Caldwell 1996) and also
373 facilitate mineral weathering of apatite (Smits et al. 2012), such as bone char. Therefore, roots

374 without AM were less successful in mining P from bone char explaining lower P acquisition in
375 comparison to plants fertilized with TSP where P was more evenly distributed throughout the
376 soil volume. In order to optimize plant P nutrition from bone char, it is important to consider
377 plant foraging ability and AM inoculation.

378 When receiving P fertilizer, root branching appeared to be a strategy to enhance AM
379 infection rather than P acquisition by roots. In comparison to plants with only root hairs, maize
380 roots associated with AM had a higher number of first-order root tips with additions of either
381 bone char or TSP. Previous studies have also reported increased branching in response to AM
382 inoculation (Berta et al. 1993; Kaldorf and Ludwig-Muller 2000; Akiyama et al. 2005). Larger
383 diameter and longer laterals in response to AM were not detected here, but have been observed
384 in other studies (Mosse 1962; Vierheilig 2004).

385 On the other hand, co-pyrolysis of wood with bone led to no change in number of root
386 tips and reduced plant P uptake in maize inoculated with AM. Yet AM colonization was
387 unaffected by biochar additions in comparison to bone char (Fig. 3b), suggesting that biochar
388 effects on plant P uptake are primarily chemical rather than biological in the present experiment.
389 As shown by the abiotic soil incubation, biochar can reduce soil P extractability. This could
390 explain the similar trends in number of root tips shown with biochar or no additions.
391 Nevertheless, biological mechanisms for a reduction in root branching under biochar application
392 cannot be excluded. Other studies have suggested that biochar may interfere with AM-plant
393 communication through sorption of signaling and allelopathic compounds (Warnock et al. 2007)
394 leading to a reduction in root thickening and specific root length of plants colonized by AM
395 (Vanek and Lehmann 2015). In these cases, biochar also demonstrated a positive effect on AM

396 colonization (LeCroy et al. 2013; Vanek and Lehmann 2015). Yet biochar-induced changes in
397 root morphology were not observed in this study (Supplementary Online Fig. S3).

398

399 **Conclusion**

400 Bone char proved to be as effective for P fertilization as commercial TSP fertilizer in a highly P-
401 fixing soil. Simultaneously managing soil P inputs and AM-root interactions improved plant
402 nutritional status. However, co-pyrolyzing maple wood with bones decreased P availability in
403 both the abiotic incubation and pot experiment. This suggests that chemical mechanisms were
404 partly responsible for biochar-induced changes in plant-available P. Along with a reduction in P
405 acquisition by maize, co-pyrolyzed maple biochar also decreased the number of root tips when
406 inoculated with AM, despite greater extractability using Olsen-P soil test and no change in AM
407 colonization. The mechanisms by which biochar affects root branching and P acquisition of AM-
408 inoculated plants warrant further research. In addition, future greenhouse experiments and field
409 studies should focus on the fertilizer efficacy of bone char with different pyrolysis temperatures
410 in other soil environments and over longer periods of time.

411

412 **Acknowledgements**

413 We are grateful for support from the Towards Sustainability Foundation, CARE-Cornell Impact
414 through Innovations Fund, McKnight Foundation, Bradfield Award, Fulbright and Huygens
415 Talent Scholarship Program. We would also like to thank Cornell Center for Materials Research
416 for help with X-ray Diffraction Analysis under NSF award number DMR-0520404, Berhanu
417 Belay and Gebermedihin Ambaw for support in procuring the soil, and Dawit Solomon for help
418 with data interpretation.

419

420 **Conflict of Interest**

421 The authors declare that they have no conflict of interest.

422

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574 **Tables**575 **Table 1** Chemical characteristics of bone char, biochar and Triple Superphosphate.

FD	T (°C)	pH	Formic -P (mg g⁻¹)	Water -P (mg g⁻¹)	C (mg g ⁻¹)	N (mg g ⁻¹)	O (mg g ⁻¹)	H (mg g ⁻¹)	P (mg g ⁻¹)	Ca (mg g ⁻¹)	Mg (mg g ⁻¹)	K (mg g ⁻¹)	Na (mg g ⁻¹)	S (mg g ⁻¹)	Fe (mg g ⁻¹)	Al (mg g ⁻¹)
RB	60	7.3 ± 0.0	79.2 ± 2.5	1.7 ± 0.1	236.4 ± 3.8	58.0 ± 1.1	197.4 ± 1.7	35.8 ± 0.6	85.9 ± 4.7	183.4 ± 10.5	3.6 ± 0.2	2.0 ± 0.3	4.3 ± 0.1	1.3 ± 0.1	0.2 ± 0.1	0.1 ± 0.0
RB	350	7.5 ± 0.0	119.6 ± 7.0	0.4 ± 0.0	180.0 ± 3.7	33.0 ± 0.5	163.4 ± 7.5	15.7 ± 0.5	127.1 ± 1.3	270.6 ± 7.2	5.2 ± 0.1	2.5 ± 0.3	5.9 ± 0.1	1.6 ± 0.0	0.1 ± 0.0	0.2 ± 0.0
RB	750	10.2 ± 0.1	147.0 ± 4.7	0.5 ± 0.0	82.1 ± 2.8	15.07 ± 0.1	104.5 ± 1.8	3.5 ± 0.1	153.2 ± 2.2	337.1 ± 8.2	5.9 ± 0.1	3.0 ± 0.4	7.3 ± 0.2	1.0 ± 0.0	0.9 ± 0.0	0.2 ± 0.0
RBW	350	7.2 ± 0.1	102.1 ± 3.3	0.7 ± 0.1	325.0 ± 4.1	27.6 ± 0.6	219.7 ± 0.4	23.6 ± 1.0	102.4 ± 1.7	224.4 ± 4.2	4.5 ± 0.1	2.3 ± 0.3	4.7 ± 0.1	1.0 ± 0.1	0.1 ± 0.0	0.2 ± 0.1
RBW	750	10.1 ± 0.0	119.0 ± 5.1	0.4 ± 0.0	337.9 ± 4.1	15.3 ± 0.8	144.8 ± 11.1	6.0 ± 0.1	116.6 ± 2.6	258.3 ± 4.5	4.7 ± 0.1	2.7 ± 0.5	5.5 ± 0.1	0.6 ± 0.0	0.3 ± 0.0	0.2 ± 0.0
TSP	n/a	3.2 ± 0.0	194.0 ± 4.1	260.5 ± 60.5	2.1 ± 0.2	0.5 ± 0.1	228.7 ± 10.5	22.1 ± 0.2	205.25 ± 1.9	161.0 ± 5.3	6.0 ± 0.1	1.5 ± 0.3	3.8 ± 0.2	10.8 ± 1.3	1.2 ± 0.0	1.0 ± 0.0
W	60	5.2 ± 0.1	0.0 ± 0.0	0.1 ± 0.0	489.9 ± 3.4	1.3 ± 0.1	439.9 ± 7.6	58.7 ± 1.2	0.06 ± 0.0	1.1 ± 0.0	0.1 ± 0.0	0.5 ± 0.0	0.1 ± 0.0	0.1 ± 0.0	0.0 ± 0.0	0.0 ± 0.0
W	350	7.0 ± 0.1	0.1 ± 0.0	0.1 ± 0.0	706.8 ± 5.4	3.4 ± 0.1	216.2 ± 1.3	40.0 ± 0.4	0.13 ± 0.0	1.1 ± 0.1	0.4 ± 0.0	1.5 ± 0.1	0.2 ± 0.0	0.1 ± 0.0	0.1 ± 0.0	0.1 ± 0.0
W	750	10.2 ± 0.0	0.3 ± 0.0	0.1 ± 0.0	928.4 ± 1.9	2.9 ± 0.1	42.9 ± 0.7	9.8 ± 0.1	0.28 ± 0.1	4.8 ± 0.1	0.6 ± 0.0	2.4 ± 0.1	0.4 ± 0.1	0.2 ± 0.0	0.1 ± 0.0	0.1 ± 0.0

576 Chemical characterization is described in more detail in Zwetsloot et al. 2015 (means and standard deviation; n=3)

577 *FD* Feedstock, *T* temperature, *RB* rendered bone, *RBW* rendered bone and wood biomass, *TSP* Triple Superphosphate, *W* maple wood.

578

579

580 **Figure Captions**

581 **Fig. 1** Olsen-extractable P (a) and resin-extractable P (b) from P-fixing soil after a five-week
582 incubation with different P sources and wood additions oven-dried at 60°C or pyrolyzed at
583 350°C or 750°C. Treatments include rendered bone meal (RB), rendered bone meal and wood
584 individually added after pyrolysis (RB+W), rendered bone meal and wood mixed before
585 pyrolysis (RBW), Triple Super Phosphate (TSP) fertilizer, and TSP and wood biochar
586 individually added (TSP+W) (n=5; LSD is the least significant difference at $p<0.05$).

587
588 **Fig. 2** Biomass and nutrient uptake of *Z. mays* after five weeks of growth fertilized with bone
589 char produced at 750°C (RB750), bone char-wood biochar co-pyrolyzed at 750°C (RBW750),
590 Triple Superphosphate (TSP), and no P additions (0-P) means and standard error are given (n=5
591 except for -RH with n=2-3): (a) plant P uptake, and (b) total plant Ca uptake, (c) total plant
592 biomass, and (d) root:shoot ratio. On the x-axis, foraging strategy of maize is altered through the
593 absence of root hairs (-RH), the presence of root hairs (+RH) and the presence of root hairs and
594 addition of AM inoculants (+RH +AM). Different capital letters indicate significant differences
595 ($p<0.05$) between treatments as determined by student's t-test.

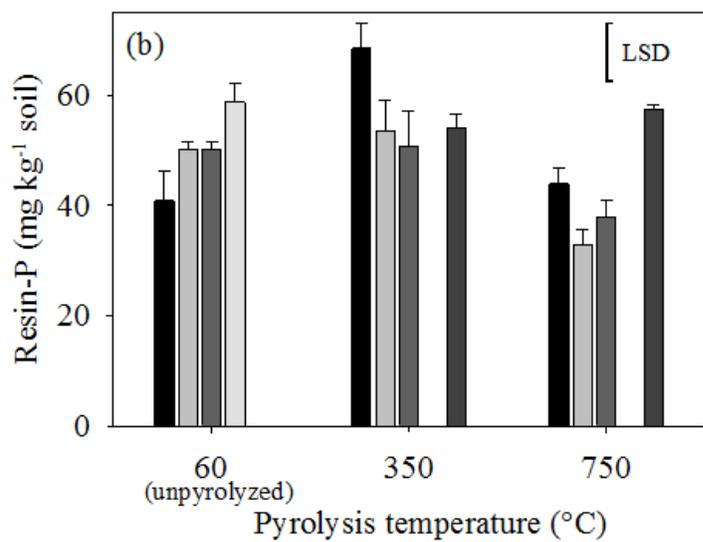
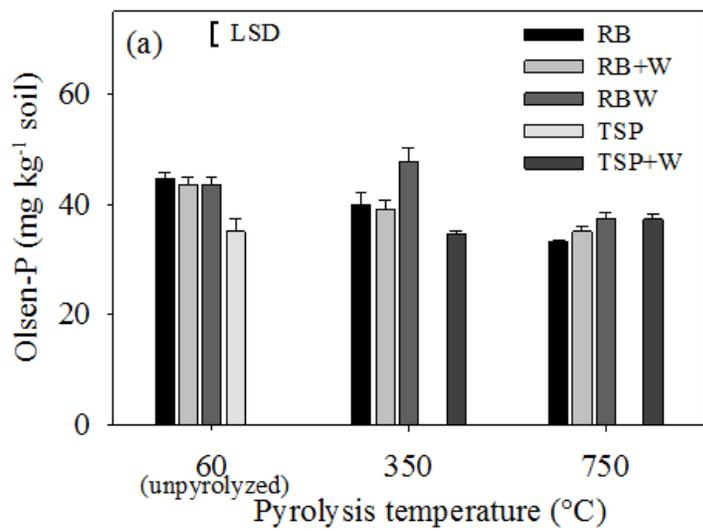
596
597 **Fig. 3** (a) Proportion of AM colonization of roots (means and standard error, n = 5, except for -
598 RH treatments where n = 2-3) and (b) Number of root tips (means and standard error, n = 4
599 except for -RH treatments where n = 2-3). Phosphorus sources include no P application (0-P),
600 rendered bone char produced at 750°C (RB750), rendered bone char-wood biochar co-pyrolyzed
601 at 750°C (RBW750) and Triple Super Phosphate (TSP). On the x-axis, rooting strategy of maize
602 is altered through the absence of root hairs (-RH), the presence of root hairs (+RH) and the

603 presence of root hairs and AM inoculants (+RH +AM). Different capital letters indicate
604 significant differences ($p < 0.05$) between treatments as determined by student's t-test.

605

606 **Figures**

607 Figure 1

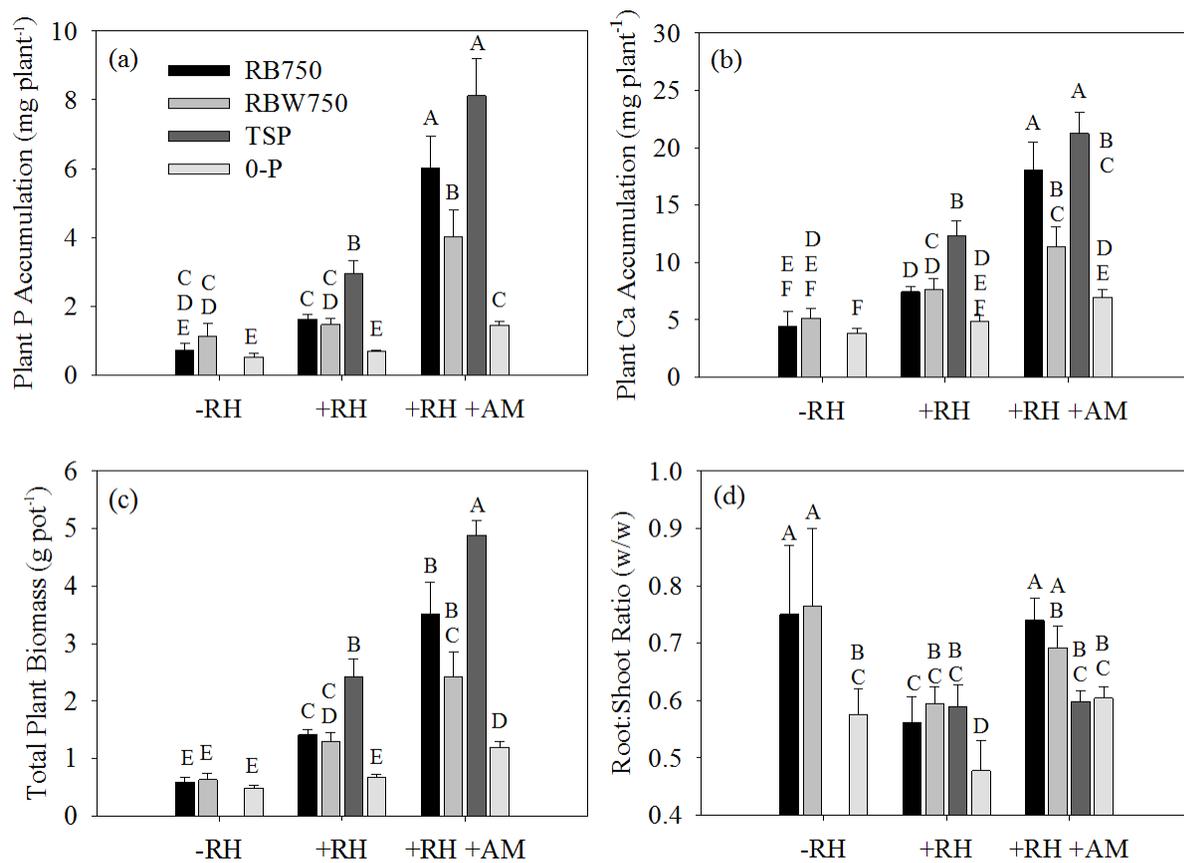


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611 Figure 2



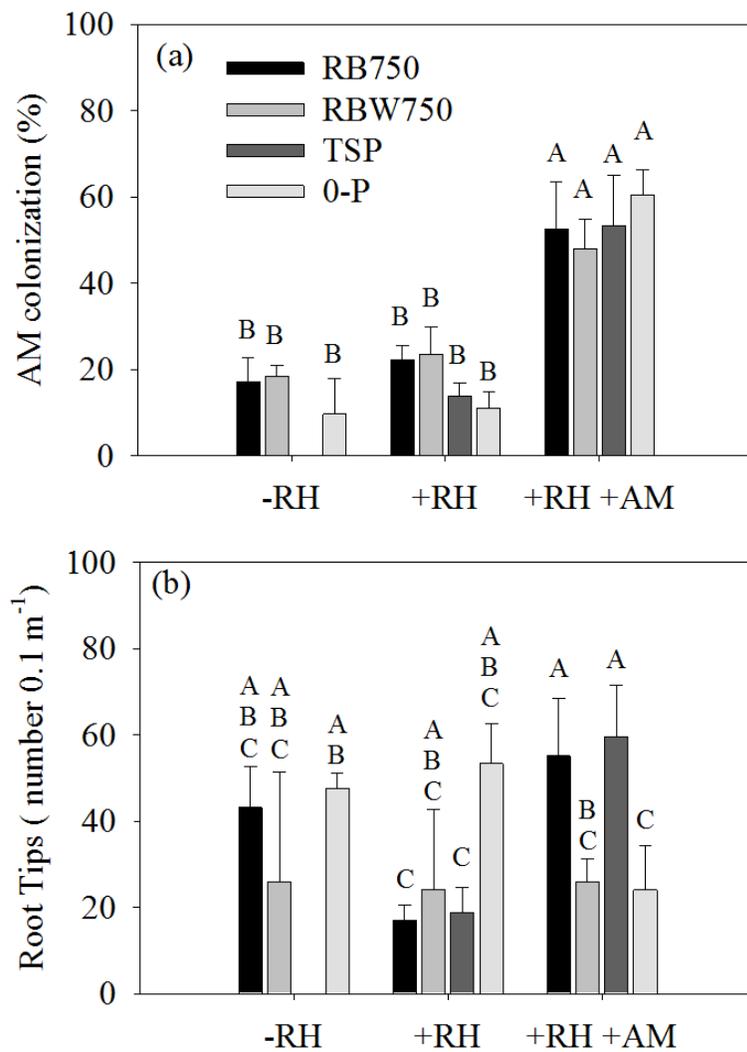
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615 Figure 3

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